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shown in Fig. 1.1. Determine the expressions of the two
second derivatives of the volume of this plot. (note: the
principle curvatures of the surface are proportional to these
second derivatives).

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various elements and compounds are $C_p = a + bT + cT^2 + dT^3 + eT^4 + fT^5 + gT^6 + hT^7 + iT^8 + jT^9 + kT^{10}$
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Thus, the thermodynamic state of a system is uniquely fixed when a small number of macroscopic, independent variables are fixed. For example, consider a gas or a liquid of constant

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composition such as a pure gas or liquid. The three key variables are pressure, P , temperature, T , and volume, V .

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